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<p>Chemical Equilibrium Questions and Answers Study.com 10^-3 and 10^3. Chemical equilibrium (definition) The state reached when the concentration of reactants and products remain constant over time. What is the general, reversible reaction for equilibrium? (With letters) aA + bB (yields) cC + dD. Equilibrium equation (general with letters) Kc = [C]^c[D]^d/[A]^a[B]^b.</p>

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<p>Chapter 17 — Study Guide — Answers 26 D5 Chemical equilibrium is said to be dynamic because A. the reaction proceeds quickly. B. the mass of the reactants is decreasing. C. the macroscopic properties are constant. D. both forward and reverse reactions are occurring. 27 D5 Equilibrium is a dynamic process because the A. macroscopic properties are not changing.</p>

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<p>Equilibrium/Acids and Bases Study Guide This model should show the three states that occur in the chemical equilibrium process. As students work, walk around to offer suggestions and guide. Have students:</p>
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<p>Chemical Equilibrium Study Guide Answers Chemical Equilibrium Constant: A reversible chemical reaction system can proceed in both directions from a set of initial conditions (concentrations) towards a state of chemical equilibrium.</p>

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Sundar Nathan received a Bachelor's degree in Electrical Engineering from Anna University, Chennai, India and a Masters degree in Biomedical Engineering from the University of Texas at Austin. Working for over a year with a team of talented Phds, MPhils and MScs from all over the world, Sundar compiled this comprehensive study guide to help students prepare diligently, understand the concepts and Crush the AP Bio Test!

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Previously published as: Chemistry: the easy way by Joseph A. Mascetta in 2019.

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